IN THE CLAIMS

Please delete all prior lists of claims in the application and insert the following list of claims:

- 1. (PREVIOUSLY PRESENTED) A method of producing hydrogen comprising: reacting water and a water-soluble oxygenated hydrocarbon having at least two carbon atoms, at a temperature not greater than about 450°C, in the presence of a metal-containing catalyst, and in the absence of added oxygen, wherein the catalyst comprises a metal selected from the group consisting of Group VIII transitional metals, alloys thereof, and mixtures thereof.
- 2. (ORIGINAL) The method of Claim 1, wherein the water and the oxygenated hydrocarbon are reacted at a temperature of from about 100°C to about 450°C, and at a pressure where the water and the oxygenated hydrocarbon are gaseous.
- 3. (ORIGINAL) The method of Claim 1, wherein the water and the oxygenated hydrocarbon are reacted at a temperature of from about 100°C to about 300°C, and at a pressure where the water and the oxygenated hydrocarbon are gaseous.
- 4. (ORIGINAL) The method of Claim 1, wherein the water and the oxygenated hydrocarbon are reacted at a temperature not greater than about 400°C, at a pressure where the water and the oxygenated hydrocarbon remain condensed liquids.
- 5. (ORIGINAL) The method of Claim 1, wherein the water and the oxygenated hydrocarbon are reacted at a pH of from about 4.0 to about 10.0.
- 6. (ORIGINAL) The method of Claim 1, wherein the catalyst comprises a metal selected from the group consisting of nickel, palladium, platinum, ruthenium, rhodium, iridium, alloys thereof, and mixtures thereof.

- 7. (ORIGINAL) The method of Claim 1, wherein the catalyst is further alloyed or mixed with a metal selected from the group consisting of Group IB metals, Group IIB metals, and Group VIIb metals.
- 8. (ORIGINAL) The method of Claim 1, wherein the catalyst is further alloyed or mixed with a metal selected from the group consisting of copper, zinc, and rhenium.
- 9. (ORIGINAL) The method of Claim 1, wherein the catalyst is adhered to a support.
- 10. (ORIGINAL) The method of Claim 9, wherein the support is selected from the group consisting of silica, alumina, zirconia, titania, ceria, carbon, silica-alumina, silica nitride, and boron nitride.
- 11. (ORIGINAL) The method of Claim 9, wherein the support is surface-modified to remove surface moieties selected from the group consisting of hydrogen and hydroxyl.
- 12. (ORIGINAL) The method of Claim 9, wherein the support is modified by treating it with a modifier selected from the group consisting of silanes, alkali compounds, and alkali earth compounds.
- 13. (ORIGINAL) The method of Claim 9, wherein the support is silica modified with trimethylethoxysilane.
 - 14. (ORIGINAL) The method of Claim 9, wherein the support is a zeolite.
 - 15-16. (CANCELED)

- 17. (PREVIOUSLY PRESENTED) The method of Claim 1, wherein the water and the oxygenated hydrocarbon are reacted at a temperature not greater than about 400°C, and further comprising reacting the water and the water-soluble oxygenated hydrocarbon in the presence of a water-soluble salt of an alkali or alkali earth metal.
- 18. (ORIGINAL) The method of Claim 17, wherein the water-soluble salt is selected from the group consisting of an alkali or an alkali earth metal hydroxide, carbonate, nitrate, or chloride salt.
- 19. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon has a carbon-to-oxygen ratio of 1:1.
- 20. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon has from 2 to 12 carbon atoms.
- 21. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon is selected from the group consisting of ethanediol, ethanedione, glycerol, glyceroldehyde, aldotetroses, aldopentoses, aldohexoses, ketotetroses, ketopentoses, ketohexoses, and alditols.
- 22. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon is selected from the group consisting of aldohexoses and corresponding alditols.
- 23. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon is selected from the group consisting of glucose and sorbitol.

24. (ORIGINAL) The method of Claim 1, wherein the water-soluble oxygenated hydrocarbon is sucrose.

25-44. (CANCELED)

- 45. (PREVIOUSLY PRESENTED) A method of producing hydrogen comprising: reacting water and a water-soluble oxygenated hydrocarbon having at least two carbon atoms, at a temperature of from about 100°C to about 450°C, and at a pressure where the water and the oxygenated hydrocarbon are gaseous, in the presence of a metal-containing catalyst, and in the absence of added oxygen, wherein the catalyst comprises a metal selected from the group consisting of Group VIII transitional metals, alloys thereof, and mixtures thereof, the catalyst being adhered to a support.
- 46. (ORIGINAL) The method of Claim 45, wherein the support is selected from the group consisting of silica, alumina, zirconia, titania, ceria, carbon, silica-alumina, silica nitride, and boron nitride, modified to remove surface moieties selected from the group consisting of hydrogen and hydroxyl.
- 47. (ORIGINAL) The method of Claim 46, wherein the support is modified by treating it with a modifier selected from the group consisting of silanes, alkali compounds, and alkali earth compounds.
- 48. (ORIGINAL) The method of Claim 45, wherein the support is silica modified with trimethylethoxysilane.
- 49. (ORIGINAL) The method of Claim 45, wherein the water-soluble oxygenated hydrocarbon has a carbon-to-oxygen ratio of 1:1.

50-56. (CANCELED)